

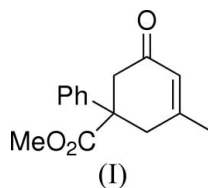
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Key indicators

Single-crystal X-ray study
 $T = 298$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.045
 wR factor = 0.122
Data-to-parameter ratio = 17.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**(*R,S*)-Methyl 3-methyl-5-oxo-1-phenylcyclohex-3-ene-1-carboxylate**The title compound, $\text{C}_{15}\text{H}_{16}\text{O}_3$, has two molecules in the asymmetric unit. The cyclohex-2-enone ring has an envelope conformation in both molecules. The crystal exhibits twinning. The molecular packing in the crystal structure is determined by weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ interactions.Received 11 May 2006
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Comment

Derivatives of the highly reactive unsaturated compounds with the cyclohex-2-enone ring can be used as precursors in the syntheses of various compounds such as vitamin E, amino acids, terpenes and others (Hu *et al.*, 2003). In addition, cyclohex-2-enone derivatives have shown a wide range of biological activities, such as antimicrobial (Li & Strobel, 2001), protecting cerebral neurocytes (Luu *et al.*, 2004) and many others. We are interested in their pharmaceutical properties. In this paper, we present the X-ray crystal structure analysis of the title compound, (I) (Fig. 1). The crystal packing is defined by weak hydrogen bonds [$\text{C}27-\text{H}271 = 0.97$ Å, $\text{H}271\cdots\text{O}2^i = 2.617$ Å, $\text{C}27\cdots\text{O}2^i = 3.204$ (2) Å, $\text{C}27-\text{H}271\cdots\text{O}2^i = 119^\circ$ where (i): $1-x, 1-y, 2-z$; Fig. 2].

Experimental

A solution of 4-carboxymethyl-4-phenyl-2,6-heptanedione (262 mg, 1 mmol) and sodium methoxide (54 mg, 1 mmol) in methanol (10 ml) was heated at 323 K for 4 h. The reaction mixture was acidified with dilute aqueous HCl, then concentrated and partitioned between water and dichloromethane. The pure product was obtained through silica gel chromatography, and diffraction quality crystals were obtained by slow evaporation of a diethyl ether/hexane (1.1) solution at room temperature.

Crystal data

 $\text{C}_{15}\text{H}_{16}\text{O}_3$
 $M_r = 244.29$
Triclinic, $P\bar{1}$
 $a = 9.317$ (4) Å
 $b = 12.332$ (5) Å
 $c = 12.734$ (5) Å
 $\alpha = 67.495$ (16) $^\circ$
 $\beta = 84.826$ (16) $^\circ$
 $\gamma = 74.278$ (16) $^\circ$ $V = 1300.9$ (9) Å³
 $Z = 4$
 $D_x = 1.247$ Mg m⁻³
Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 298$ (1) K
Block, colourless
 $0.27 \times 0.26 \times 0.17$ mm

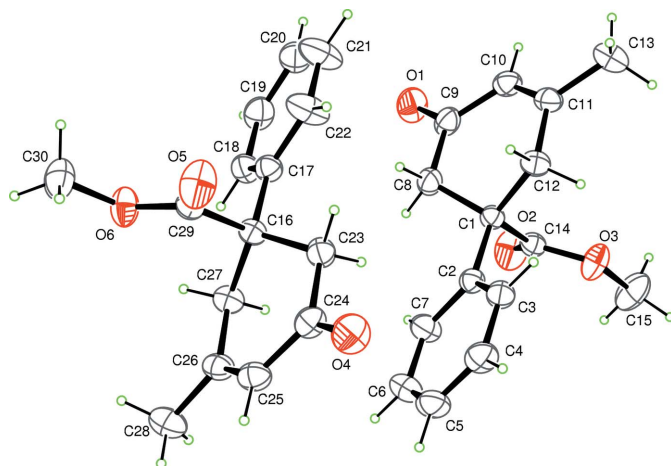


Figure 1
The asymmetric unit of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 40% probability level.

Data collection

Rigaku R-AXIS RAPID
diffractometer
 ω scans
Absorption correction: none
12919 measured reflections

5900 independent reflections
3491 reflections with $F^2 > 2\sigma(F^2)$
 $R_{\text{int}} = 0.023$
 $\theta_{\text{max}} = 27.5^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.046$
 $wR(F^2) = 0.122$
 $S = 1.02$
5900 reflections
336 parameters
H-atom parameters constrained

$w = 1/[0.0008F_o^2 + \sigma(F_o^2)]/(4F_o^2)$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.22 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.15 \text{ e } \text{\AA}^{-3}$
Extinction correction: Larson
(1970), equation 22
Extinction coefficient: 66 (15)

The methyl H atoms were positioned with idealized geometry and allowed to rotate but not to tip, with C—H distances of 0.96 Å, and refined using a riding model with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. All other H atoms were placed in geometrically idealized positions with C—H distances of 0.93 Å (sp^2) or 0.97 Å (sp^3), and were refined using a riding model with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2004); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *ORTEP-3 for Windows*

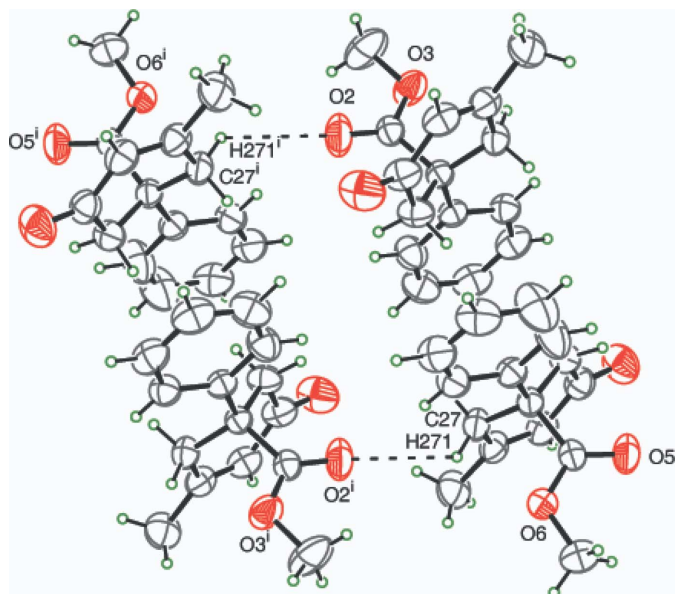


Figure 2
View of the hydrogen bonds between neighbouring molecules. Hydrogen bonds are indicated by dashed lines. Symmetry code (i): $1 - x, 1 - y, 2 - z$.

(Farrugia, 1997); software used to prepare material for publication: *CrystalStructure*.

X-ray data were collected at the analysis centre of Zhejiang University.

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